

POGORELOV, I.D., inzh.; ZEMSKOV, P.I., kand.tekhn.nauk

Cold electric-arc welding of cast iron. Izv.vys.ucheb.
zav.; mashinostr. no.2:146-152 '59. (MIRA 13:3)

1. Khar'kovskiy institut inzhenerov kommunal'nogo stroitel'-
stva i Khar'kovskiy zavod "Serp i molot".
(Electric welding)

ZEMSKOV, Pavel Ivanovich; YAKUSHINA, Yelena Nikolayevna;
KHARCHENKO, Yevgeniy Nikolayevich; ZUBENKO, I.F., dots.,
otv. red.; ALYAB'YEV, N.Z., red.

[Materials and coatings for the piston rings of motor-
vehicle and tractor engines] Materialy i pokrytiia porsh-
nevykh kolets avtotrektornykh dvigatelei. Khar'kov, Izd-
vo Khar'kovskogo univ., 1963. 129 p. (MIRA 17:8)

ZEMSKOV, P.I.; YAKUSHINA, Ye.N.

Antifriction pseudoalloys for engine bearings. Avt. prom. 29
no.7:21-23 JI '63. (MIRA 16:8)

1. Khar'kovskiy traktornyy zavod.
(Bearing metals)

ZEMSKOV, P.I., kand. tekhn. nauk; YAKUSHINA, Ye.N., inzh.

Using pseudocalloys as substitutes for bronzes and babbitts.
Mashinostroenie no.3:110 My-Je '63. (MIRA 16:7)

1. Khar'kovskiy traktornyy zavod.
(Powder metallurgy)

ZEMSKOV, P.I., inzh.; POGORELOV, I.D., inzh.; YAKUSHINA, Ye.N., inzh.;
KHARCHENKO, Ye.N., inzh.

Welding and soldering during the repair of ALLOV aluminum
alloy parts. Svar. proizv. no.8:40-41 Ag '63. (MIRA 17:1)

1. Khar'kovskiy zavod "Serp i molot".

ZEMSKOV, P.I., dotsent; POGORELOV, I.D.; YAKUSEINA, Ye.N.

Soldering aluminum parts. Mashinostroitel' no.7:38-39 JI '62.

(MIRA 15:7)

(Aluminum—Welding)

ZEMSKOV, P.I., kand.tekhn.nauk, dotsent; POGORELOV, I.D., inzh.; BALLYUK,
B.K., inzh.

Investigating the performance of engine bimetallic bushings made with
ASM alloy. Izv.vys.ucheb.zav.; mashinostr. no.11:79-83 '61.
(MIRA 14:12)

1. Khar'kovskiy institut inzhenerov kommunal'nogo stroitel'stva.
(Aluminum alloys--Testing)

ZEMSKOV, P.I.; POGORELOV, I.D.

Device for testing the hardness of shafts. Mashinostroitel'
no.5837 My '62. (MIRA 15:5)
(Rockwell test)

ZEMSKOV, P.I.; YAKUSHIN, Ye.N.; KHARCHENKO, Ye.N.

Wearing resistance of crankshafts from high-strength cast iron.
Trakt. i sel'khoz mash. no.1:41-43 Ja '64. (MIRA 17:4)

1. Khar'kovskiy traktorny zavod.

ZEMSKOV, P.I.; POGORELOV, P.D.

Devices for measuring the hardness of control rod and crankshaft
in a Rockwell press. Zav.lab. 28 no.3:366-367 '62. (MIRA 15:4)

1. Khar'kovskiy zavod "Serp i molot".
(Rockwell test)

ZEMSKOV, P.I., kand.tekhn.nauk, dotsent; POGORELOV, I.D., inzh.

Investigating high-grade cast iron as material for piston rings
of motor-vehicle engines. Izv.vys.ucheb.zav.; mashinostr. no.4:
138-147 '62. (MIRA 15:7)

1. Khar'kovskiy institut kommunal'nogo khozyaystva.
(Cast iron—Testing)
(Piston rings)

ZEMSKOV, P.I., dotsent, kand.tekhn.nauk; POGORELOV, I.D., inzh.

Making distributing shafts of gray iron. Izv. vys. ucheb. zav.;
mashinostr. no. 10:106-114 '60. (MIRA 14:1)

1. Khar'kovskiy kommunal'nyy institut.
(Tractors--Engines)

ZEMSKOV, P.I., kand.tekhn.nauk.dotsent; POGORELOV, I.D., inzh.

Comparative data on antifriction properties of some plastics. Izv.vys.
ucheb.zav.; mashinostr. no.3:77-81 '60. (MIRA 14:3)

1. Khar'kovskiy institut inzhenerov kommunal'nogo stroitel'stva.
(Plastics--Testing)

ZEMSKOV, P.I., dotsent, kand.tekhn.nauk; POGORELOV, I.D., inzh.

Experience in welding and soldering aluminum parts at the "Serp i Molot" Plant in Kharkov. Izv.vys.ucheb.zav.: mashinost'r. no.7: 89-95 '59. (MIRA 13:6)

1. Khar'kovskiy institut inzhenerov kommunal'nogo stroitel'stva.
(Electric welding) (Solder and soldering)

ZEMSKOV, P. I.

ZEMSKOV, P. I. --"Investigation of the Wear on Piston Rings of Automobile Engines." Min Higher Education Ukrainian SSR. Khar'kov, 1955.
(Dissertation for the Degree of Candidate in Technical Sciences.)

So.: Knizhnaya Litopis', No. 7, 1956.

ZEMSKOV, P.I., kand.tekhn.nauk, dotsent; POGORELOV, P.D., inzh.

Device for measuring the hardness of a crankshaft. Izv.vys.ucheb.-
zav.; mashinostr. no.7:177 '61. (MIRA 14:9)

1. Khar'kovskiy institut inzhenerov kommunal'nogo stroitel'stva.
(Hardness--Measurement)

S/145/60/000/003/008/010
D221/D301

AUTHORS: Zemskov, P.I., Candidate of Technical Sciences, Docent
and Pogorelov, I.D., Engineer

TITLE: Comparative data on antifrictional properties of
certain plastics

PERIODICAL: Izvestiya vysshikh uchbenykh zavedeniy. Mashino-
stroyeniye, no. 3, 1960, 77 - 81

TEXT: The Central Factory Laboratory at the "Serp i moloť" plant
in Khar'kov investigated antifrictional properties of plastics.
Specimen bushes from normal caprone, and also caprone with 3-5 %
of an graphite admixture as well as tekstolite, voloknite and feno-
plast were tested. The inserts worked with a steel shaft, cr-45
(st-45), R_c - 60 and H_b - 170, and a high strength cast iron, R_c -
40, H_b - 170. Bronze, babbitt and cast iron bushes were tested for
comparison. The purpose was the study of the effect of speed, load,
lubricant, material and hardness of shaft on the coefficient of
friction. Tests were carried out on an MVI (MI) machine. The wear on
Card 1/3

Comparative data on antifrictional ...

S/145/60/000/003/008/010
D221/D301

bushes was measured by analytical balance. The surface of rollers which imitated the shaft was ground. Hardened and untreated rollers were employed. The lubricant was formed by Industrial'noye (Industrial) 20, AK-10 and ДП-14 (DP-14) oils and персоль АШ (persolidol ASH) grease. The minimum wear was exhibited by caprone, especially when having 3 - 5 % graphite additive, $Q = 0.5 \mu$ and $Q = 0.1 - 0.2 \mu$ respectively. The wear for other plastics was 4 - 5 times greater. This amounted to 2.5μ for bronze and babbitt bushes and 10μ in the case of cast iron insert. The caprone with graphite indicated best antifrictional properties, but the minimum friction torque is shown by babbitt. This torque has a marked tendency to rise at the beginning of running-in of caprone and then falls again. The friction coefficient of caprone depends on load and speed, as indicated by graphs. It is highest at low loads. The wear of caprone inserts increases with the hardness of shaft. Cast iron shaft exhibits a smaller friction which is apparently due to greater porosity of former and also on account of lubricating properties of its graphite. The friction in caprone is lower when AK-10 oil is used than when lubricating with "Industrial 20". The minimum friction and wear take place during work of caprone with grease. The Card 2/3

Comparative data on antifrictional ...

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D221/D301

authors conclude that caprone is the most suitable for bushes. In the case of inadequate lubrication, caprone with 3 - 5 % of graphite addition is recommended. There are 5 figures.

ASSOCIATION: Khar'kovskiy institut inzhenerov 'komunal'nogo stroitel'stva (Khar'kov Institute of Engineers of Civil Construction)

SUBMITTED: July 17, 1959

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Card 3/3

ZEMSKOV, P. I.; POGORELOV, I. D.

Investigating the wear of piston rings made of high-strength
cast iron in motor-vehicle engines. Avt. prem. 28 no.9:30-33
S '62. (MIRA 15:10)

1. Khar'kovskiy zavod "Serp i molot".

(Piston rings—Testing)

FOGORELOV, I.D., inzh.; ZEMSKOV, P.I., inzh.

Rockwell hardness testing press for crankshaft journals.
Metalloved. i term. obr. met. no.7:57 JI '61. (MIRA 14:6)

1. Khar'kovskiy zavod "Serp i molot."
(Crank and crankshafts--Testing)
(Hardness--Testing)

AID P - 4288

Subject : USSR/Engineering

Card 1/1 Pub. 128 - 13/25

Authors : Yengalychev, I. M., Engineer and P. I. Zemskov

Title : Experience in sulphide coating at the Khar'kov plant
"Serp 1 Molot".

Periodical : Vest. mash., #2, p. 46-48, F 1956

Abstract : The sulphide process for surface coating of cast iron and steel parts to reduce wear and scuffing, particularly between piston rings and cylinders, has been investigated in the Khar'kov plant. Various caustic sulphide baths at different temperatures of treatment are reported. Photos, tables, chart.

Institution : None

Submitted : No date

ZEMSKOV, P.I., inzh.; YAKUSHINA, Ye.N., inzh.; KHARCHENKO, Ye.N., inzh.;
KHAVINA, R.B., inzh.

Engine pinions made from high-strength cast iron. Mashinostroenie
no.2:12-14 Mr-Ap '65. (MIRA 18:6)

ZEMSKOV, P.I., kandi.tekhn.nauk; KHAVINA, R.B., inzh.; DECTYAREVA, O.F., inzh.;
YAKUSHINA, Ye.N., inzh.; KHARCHENKO, Ye.N., inzh.; ANISHCHENKO, V.V.,
inzh.

Capron pinions for motor-vehicle engines. Mashinostroenie
no.6:42-44 N-D '65. (MIRA 18:12)

ZEMSKOV, P.I., dotsent; YAKUSHINA, Ye.N., inzh.; KHARCHENKO, Ye.N., inzh.;
KHAVINA, R.B., inzh.; DEGTYAREVA, O.F., inzh.

Cermet piston rings. Izv. vys. ucheb. zav.; mashinostr. no. 10:
123-128 '65 (MIRA 19:1)

1. Submitted April 17, 1964.

L 07438-67 EWT(m)/EWP(t)/ETI IJP(c) JD
 ACC NR: AP6029220 (A, N) SOURCE CODE: UR/0145/66/000/004/0059/0063
 AUTHOR: Zemskov, P. I. (Docent); Yakushina, Ye. N. (Engineer); Kharchenko, Ye. N. (Engineer); Khavina, R. B. (Engineer); Degtyareva, O. F. (Engineer)
 ORG: None
 TITLE: Improving the durability of chrome-plated piston rings
 SOURCE: IVUZ. Mashinostroyeniye, no. 4, 1966, 59-63
 TOPIC TAGS: engine piston, engine cylinder, carburization, chromium plating
 ABSTRACT: Methods are proposed for increasing the wear resistance of the upper piston rings in tractor engines by carburizing the chrome-plated surface. The surfaces of the rings and cylinder were knurled before chrome plating. The knurl depressions were tetrahedral pyramids with a base of 0.5x0.5 mm located 2 mm apart with a depth of 0.18-0.25 mm. After chrome plating, the rings were chemically heat treated in a solid carburizer of the following composition: carbon--50%, Na₂CO₃--20%, Fe (filings)--30%. The heat treatment was continued for 5 hours at 950°C. X-ray structural analysis showed a gray layer of chromium carbide on the metal surface. This layer was 60-80 μ thick and was not etched by a 3-4% solution of HNO₃ or a 15-20% solution of HCl. The carbide layer is heat- and acid-resistant with a hardness of 1400-1600 kg/cm². It was found that carburization increases the service life of chrome-plated piston rings by a factor of 1.3-2.2. The article was presented for publication by A. I. Pogorelov, Lecturer at Kharkov Municipal Engineering Institute. Orig. art. has: 3 figures. 1 table.
 SUB CODE: 10, 13/ SUBM DATE: 18Jun64/ ORIG REF: 002
 CIRD 1/1 UDC: 62-47/-242

ZEMSKOV P.M.
KAGANSKIY, M.G.; ~~ZEMSKOV~~, P.M.

Sensitive instrument for conductometric analysis. Dum.prom. 29 no.10:
21-22 0 '54. (MLRA 7:11)

1. Tsentral'nyy nauchno-issledovatel'skiy institut bumagi.
(Volumetric analysis)

ZEMSKOV, P.

Consolidation of automotive transportation units and centralization
of hauling leads to technical development. Avt. transp. 37 no.2:12-15
F '59. (MIRA 13:1)

1. Glavnyy inzhener Glavmosavtotransa.
(Transportation, Automotive)

ZEMSKOV, S.

Fulfilling the annual plan ahead of time. Avt.transp. 32 no.6:14-16
Je, '54. (MLRA 7:9)

1. Avtokolonna No. 60 Tul'skogo oblastnogo avtotresta.
(Transportation, Automotive)

AUTHORS: Sakharova, N.N., Zemskov, S.V. SOV/ 20-120-3-28/67

TITLE: Double Chlorides of the Elements of the Cerium Group
With Trimethylamine Chloride (O dvoynykh khloridakh
elementov tseriyevoy gruppy s khloridom trimetilamina)

PERIODICAL: Doklady Akademii nauk SSSR, 1958, Vol. 120, Nr 3,
pp. 539-540 (USSR)

ABSTRACT: Earlier the first author investigated double chlorides
(Ref 1) and nitrites (together with G. V. Medoks, Ref 2)
of the elements of the group in question with the corresponding
tetraphenyl-phosphonium (Refs 3-5). Later Medoks (Ref 6)
isolated double nitrates of lanthanum, praseodymium, and
neodymium with triphenyl benzyl phosphonium. In continuation
of the above mentioned papers the authors included the
methylamine-dimethylamine- and trimethylamine chlorides
in their investigations. They intended to clear up the
nature of the complex formation of the chlorides of rare
earths with the corresponding salts of the amides in
question in the presence of water and in nonaqueous media
according to their degree of substitution. As an introduction
it was stated that the chlorides of the 4 metals from several

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Double Chlorides of the Elements of the Cerium
Group With Trimethylamine Chloride

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compounds in the combination with trimethylamine chloride in the presence of 96%-ethyl alcohol, the formation of which is dependent upon the relative amounts of the substances entering interaction, and upon the crystallization conditions. As a result of the conducted work double chlorides of rare earths with trimethylamine in the presence 96%-ethanol were produced. They correspond to the general formulae $\text{MeCl}_3 \cdot 4(\text{CH}_3)_3\text{N} \cdot \text{HCl}$, where Me denotes La, Ce, Pr and Nd. The properties of these compounds were examined, and the solubility in water and organic solvents was determined. There are 6 references, 6 of which are Soviet.

ASSOCIATION: Saratovskiy gosudarstvennyy universitet im. N.G. Chernyshevskogo (Saratov State University imeni N. G. Chernyshevskiy)

PRESENTED: January 24, 1958, by I. I. Chernyayev, Member, Academy of Sciences, USSR

SUBMITTED: September 10, 1956
Card 2/3

Double Chlorides of the Elements of the Cerium
Group With Trimethylamine Chloride

SOV/20-120-3-28/67

1. Rare earth chlorides--Chemical reactions
2. Amine chlorides--Chemical reactions
3. Ethanol--Applications
4. Complex compounds--Properties
5. Organic solvents--Performance

Card 3/3

S/137/62/000/001/033/237
A060/A101

AUTHORS: Koz'min, Yu. A., Zemskov, S. V., Ryabinin, A. I.

TITLE: Application of the sulfide-sulfite method in the processing of tellurium-containing materials

PERIODICAL: Referativnyy zhurnal, Metallurgiya, no. 1, 1962, 22, abstract 10164 ("Metallurg: i khim. prom-st' Kazakhstana. Nauchno-tekhn. sb.", 1961, no. 1(11), 23-25)

TEXT: The authors studied the possibility of applying the sulfide-sulfite method to the processing of rich Te-containing products. It is shown that this method ensures the extraction of 93 - 94% of the Te from the primary hydroxide (at an Na_2S expenditure of 5 - 6 kg per 1 kg Te) as against 60 - 70% extraction by the soda method, and when soda slags are processed - 81-84% versus 40-50%. The reagent expenditure and process duration are reduced when the sulfide-sulfite method is used. ✓

G. Svodtseva

[Abstracter's note: Complete translation]

Card 1/1

ZEMSKOV, S.V.; PTITSYN, B.V. [deceased]

Oxidation of cis- and trans-isomers of bivalent platinum by
sodium vanadate. Zhur. neorg. khim. 10 no.6:1502-1503 Je '65.
(MIRA 18:6)

1. Institut neorganicheskoy khimii Sibirskogo otdeleniya AN SSSR.

ZEMSKOV, S.V.; PTITSYN, B.V.

Oxidation of the platinite series by sodium vanadate. Dokl. AN SSSR
160 no.2:343-345 Ja '65. (MIRA 18:2)

1. Institut neorganicheskoy khimii Sibirskogo otdeleniya AN SSSR.
2. Chlen-korrespondent AN SSSR (for Ptitsyn).

SAKHAROVA, N.M.; ZEMSKOV, S.V.

Double chlorides of cerium group elements and trimethylamine
chloride. Dokl. AN SSSR 120 no. 3:539-540 My '58. (MIRA 11:7)

1. Saratovskiy gosudarstvennyy universitet im. N.G.Chernyshevskogo.
Predstavleno akademikom I.I.Chernyayevym.
(Rare earth chlorides)
(Trimethylamine)

S/137/62/000/001/032/237
AO60/A101

AUTHORS: Koz'min, Yu. A., Ryabinin, A. I., Zemskov, S. V.

TITLE: On the oxidation of tellurium up to the tetravalent state

PERIODICAL: Referativnyy zhurnal, Metallurgiya, no. 1, 1962, 22, abstract 10163
("Metallurg. i khim. prom-st' Kazakhstana. Nauchn.-tekhn. sb.",
1961, no. 2(12), 57-61)

TEXT: A study was made as to the possibility of obtaining water-soluble Te from anodic copper electrolytic slimes by producing a definite composition of the gaseous phase and the charge preparation schedule. It was established that in the course of oxidizing roasting of the slime with soda the Te is transformed almost entirely into the hexa-valent variety, and in the course of the aqueous lixiviation of the clinker if remains in the cake. Calcination of the clinker in a stream of CO₂ or N₂ at 700 - 750°C affords the possibility of transforming 70% and more of the Te into the tetra-valent, soluble variety. The reduction of Te to Te⁺⁺⁺ by carbon monoxide occurs at lower temperatures. In the laboratory investigations the transformation of Te into Te⁺⁺⁺ constituted 80 - 90%.
G. Svodtseva

[Abstracter's note: Complete translation]

Card 1/1

NIVOSELOV, R.I.; ZEMSKOV, S.V.; PTITSYN, B.V.

Oxidation of tetrammine platinum chloride by iodine. Dokl. AN SSSR 158
no.5:1133-1135 0 '64. (MIRA 17:10)

1. Institut neorganicheskoy khimii Sibirskogo otdeleniya AN SSSR. 2.
Chlen-korrespondent AN SSSR (for Ptitsyn).

CHERNYAYEV, I.I.; ZEMSKOV, S.V.; PTITSYN, B.V. [deceased]

Oxidation-reduction properties of nitrite complexes of platinum.
Zhur.neorg.khim. 10 no.11:2404-2410 N '65.

(MIRA 18:12)

1. Institut neorganicheskoy khimii Sibirskogo otdeleniya AN SSSR.
Submitted May 5, 1964.

ZEMSKOV, V.A. (Leningrad); MAKAR'YEV, B.M. (Leningrad)

Improvement of the characteristics of automatic control systems of
a certain class using nonlinear control signs. Izv. AN SSSR.
Tekh. kib. no.6:60-71 N-D '63. (MIRA 17:4)

ZEMSKOV, V.A. (Leningrad); MAKAR'YEV, B.M. (Leningrad)

Improvement of the interference rejection of a class of automatic control systems by means of nonlinear control laws. Izv. AN SSSR. Tekh. kib. no.1:143-149 Ja-F '65.

(MIRA 18:4)

GONCHAREVICH, Igor' Fomich; ~~ZEMSKOV~~, Vasilii Dmitriyevich; KORESHKOV,
Viktor Ivanovich; BRILLIANTOV, V.V., otv.red.; GARBOR, P.N.,
red.izd-va; BOLDYREVA, Z.A., tekhn.red.

[Shaker screens and conveyers] Vibratsionnye grokhoty i
konveieri. Moskva, Gos.nauchno-tekhn.izd-vo lit-ry po gornomu
delu, 1960. 214 p. (MIRA 14:3)
(Screens (Mining)) (Conveying machinery)

Z E M S K O V , V. D.

ANDRUKYEV, S. Ye.; BOKIY, B. V.; GORODETSKIY, P. I.; GREYVER, N. S.; SHCHUKIN, A. A.
GERONT'YEV, V. I.; SKOCHINSKIY, A. A.; TERFIGOREV, A. M.; SHEVYAKOV, L. D.;
SPIVAKOVSKIY, A. A.; VERKHOVSKIY, I. M.; VORONKOV, I. M.; YELANCHIK, G. M.;
KASHIN, N. V.; SLOBODKIN, M. I.; GUZENKOV, P. G.; ZEMSKOV, V. D.; NOVIKOV, F. S.
OSIETSKIY, V. M.; SOSUNOV, G. I.; YASYUKOVICH, S. M.; KHAN, G. A.; POPOV, V. M.

In memory of Professor Levenson. Gor.zhur. no.9:60 S '55.
(MIRA 8:8)

(Levenson, Lev Borisovich, 1878-1955)

ZEMSKOV, V.D.; KOVAL', P.V.

Calculating the impact mechanism in dynamic coal plows.

Nauch. trudy MGI no.21:201-214 '57.

(MIRA 11:9)

(Coal mining machinery)

ZEMSKOV, V.D., kandidat tekhnicheskikh nauk; BRILLIANTOV, V.V., kandidat tekhnicheskikh nauk; VINOGRADOV, N.N., kandidat tekhnicheskikh nauk.

A misconception constantly appearing in books on ore dressing.
Ugol' 32 no.4:47-48 Ap '57. (MLRA 10:5)

1. Moskovskiy gornyy institut.
(Ore dressing)

ZEMSKOV V. D.

127-58-1-15/28

AUTHORS: Verkhovskiy, I.M., Professor, Doctor of Technical Sciences;
Zemskov, V.D., and Vinogradov, N.N., Candidates of Technic-
al Sciences; Arutinov, O.M., Engineer-Physicist

TITLE: Investigation by the Gamma-Location Method of Some Re-
gularities in the Jigging Process (Issledovaniye nekotorykh
zakonomernostey protsessa otsadki metodom gamma-lokatsii)

PERIODICAL: Gornyy Zhurnal, 1958, Nr 1, pp 53-56 (USSR)

ABSTRACT: Numerous versions of the "marked atoms" method employed at
present do not make possible the study of the kinetics of
the spatial motion of grains in media with different ab-
sorption coefficients. Therefore, the Chair of Concentra-
tion at the Moscow Mining Institute developed a new method,
named "gamma-location", for the continuous study of mineral
grain motion. The gamma-location method makes it possible
to record continuously the spatial displacements of a part-
icle under investigation into which a radioactive isotope
with hard gamma-radiation is inserted. The block-diagram
of the equipment is shown in Figure 1. The equipment con-
sists of the following basic parts: 1) an indicator

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Investigation by the Gamma-Location Method of Some Regularities in the Jigging Process

system; 2) an electronic computer; 3) an oscillograph; 4) a high-voltage stabilized amplifier, and 5) a stabilizer of voltage. The essence of the gamma-location method consists in that the spatial motion of the grains is broken-down into 3 coordinates, by means of special systems of indicators. The motion of the grain along each coordinate is continuously recorded as a function of time. The indicator systems consist of standard discharge counters situated according to a definite system, called "carpets". The electronic computer converts the pulses into direct current proportional to their frequency. It consists of several similar channels corresponding to the number of coordinates being measured simultaneously. The results permit the discovery of basic laws of grain motion in jigging machines. For instance, the graph of the motion of a heavy grain represents a monotonous function (shown in Figure 2) which indicates that the grain continuously sinks to the bottom layer. The motion of light grains has a "jump-like" character shown in Figure 3. Analysis of the results shows that the state of intermediate, slightly loosened layers

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127-58-1-15/28

Investigation by the Gamma-Location Method of Some Regularities in the Jigging Process

(thickness and degree of loosening) is a criterion of the jigging process which determines the specific weight of separation and the quality of concentration products. The intermediate layers serve as a filter which passes down only grains of a definite specific gravity. The filtering properties of the intermediate layer depend upon many factors: specific gravity and size of the grains, the magnitude of hydrodynamic forces, and mechanical interaction forces between the grains. The establishment of the effects of these factors calls for extensive investigations. The article contains 1 figure, 2 graphs, and 7 references, of which 5 are Soviet and 2 English.

ASSOCIATION: Moskovskiy gornyy institut (Moscow Mining Institute)

AVAILABLE: Library of Congress

Card 3/3 1. Tracers-Applications 2. Isotopes (Radioactive)-Applications
 3. Gamma rays-Applications 4. Mining engineering-USSR

ZEMSKOV, V.D., dots.

Basic problems in the theory of unbalanced shaker screen performance. Nauch. dokl. vys. shkoly; gor. delo no.2:281-296 '58.
(MIRA 11:6)

1. Predstavlena kafedroy obogashcheniya poleznykh iskopayemykh Moskovskogo gornogo instituta im. I.V. Stalina.
(Screens (Mining))

ZEIMSKOV, V.D., dots., kand.tekhn.nauk

Analysis of transition processes in vibrating screens. Nauch.dokl.
vys.shkoly; gor.delo. no.4:231-240 '58. (MIRA 12:1)

1. Predstavleno kafedroy obogashcheniya poleznykh iskopayemykh Moskov-
skogo gornogo instituta imeni I.V. Stalina.
(Screens (Mining))

ZEMSKOV, V.D. BRILLIANTOV, V.V.; VINOGRADOV, N.N.; SHOKHIN, V.N.
VLAD, P.

Electric measurement methods in investigating wet gravity ore
dressing processes. Nauch. trudy MGI no. 32:5-14 '60.

(MIRA 14:2)

(Ore dressing)

(Electric measurements)

ZHGULEV, Al'bert Sergeyevich; ZEMSKOV, V.D., kand. tekhn. nauk,
retsensent; BRILLIANTOV, V.V., kand. tekhn. nauk, otv.
red.; GARBER, T.N., red.izd-va; LAVRENT'YEVA, L.G.,
tekhn. red.

[Electrovibrating machine operator] Mashinist elektrovib-
ratsionnoi mashiny. Moskva, Gosgortekhzdat, 1963. 91 p.

(MIRA 16:12)

(Ore dressing--Equipment and supplies) (Vibrators)

ZEMSKOV, V.I.

Design of small bridges and culverts in regions with ever frozen
subsoils. Avt.dor. 25 no.1:24-25 Ja '62. (MIRA 15:2)
(Russia, Northern--Bridges--Design) (Russia, Northern--Culverts)

ZEMSKOV, V.M., prof.

C-reactive protein in the diagnosis of some cardiovascular diseases
(coronary insufficiency, rheumatism). Vrach. delo no.6:22-26 Ja '61.
(MIRA 15:1)

1. Kafedra fakul'tetskoy terapii (zaveduyushchiy - M.N.Tumanovskiy)
i kafedra mikrobiologii (zaveduyushchiy - prof. M.V.Zemskov)
Voronezhskogo meditsinskogo instituta.
(BLOOD PROTEINS) (HEART FAILURE)
(RHEUMATIC FEVER)

ZEMSKOV, V.M.

Dynamics of C-reactive protein, erythrocyte sedimentation reaction,
and leucocytes in myocardial infarct. Lab. delo 7 no.3:20-22.
Mr '61. (MIRA 14:3)

1. Kafedra fakul'tetskoy terapii (zav. - prof. M.N.Tumanovskiy)
i Kafedra mikrobiologii (zav. - prof. M.V.Zemskov) Voronezhskogo
meditsinskogo instituta.

(BLOOD PROTEINS)

(LEUCOCYTES)

(ERYTHROCYTES)

(HEART—INFARCTION)

ZEMSKOV, V.M. (Moskva)

Methodology of studying the plasmocyte reaction in immune animals.
Lab. delo 10 no.5:270-272 '64. (MIRA 17:5)

PETROV, R.V.; ZEMSKOV, V.M.; PASHININ, P.M.

Place of the formation of S-reactive protein in radiation
injury. Radiobiologia 5 no.4:511-513 '65. (MIRA 18:9)

ZEMSKOV, Vladimir Mikhaylovich; SOROKO, Ya.I., red.

[Weapons of invisible enemies] Ouzhie nevidimyykh vragov.
Moskva, Izd-vo "Znanie," 1964. 39 p. (Novoe v zhizni,
nauke, tekhnike. VIII Seriya: Biologiya i meditsina, no.11)
(MIRA 17:7)

ZEMSKOV, V.M.

Morphological indices of immunity; a review. Zhur. mikrobiol.,
epid. i imm. 41 no. 2:25-28 F '64. (MIRA 17:9)

137-58-6-12122

Translation from: Referativnyy zhurnal, Metallurgiya, 1958, Nr 6, p 134 (USSR)

AUTHORS: Petrov, D.A., Kekua, M.G., Dashevskiy, M.Ya., Zemskov,
V.S., Petrusevich, R.L.

TITLE: Progress of Work on the Refining of Germanium by Means of
Crystallization and Achievement of Germanium Single Crystals
With Longitudinally Homogeneous Properties (Razvitiye rabot
po ochistke germaniya metodami kristallizatsii i polucheniye
monokristallov germaniya s ravnomernymi svoystvami po
dline)

PERIODICAL: V sb.: Vopr. metallurgii i fiz. poluprovodnikov. Moscow,
AN SSSR, 1957, pp 50-58

ABSTRACT: Experiments were performed in order to investigate the
possibilities of producing single crystals with uniform longi-
tudinal and cross-sectional distribution of impurities by means
of pulling at a controlled rate as well as by pulling accompanied
by constant feeding of pure Ge into the melt. The raw poly-
crystalline material with a ρ of 5-20 ohm/cm was purified by
means of zonal recrystallization under a vacuum of 10^{-4} mm
Hg, or in a stream of H_2 , until it exhibited a ρ of 50-60
ohm/cm. A high-frequency heating apparatus with three

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137-58-6-12122

Progress of Work on the Refining of Germanium (cont.)

induction units moving back and forth was employed. It was found that single Ge crystals, grown by the Chokhralskiy method from purified material with a ρ of 50-60 ohm/cm in conjunction with mixing of the melt by rotating the crucible and the growing crystal, can achieve ρ and τ values of 60 ohm/cm and 1000 μ sec, respectively; if the rotation is omitted from the growth process, the ρ and the τ amount to 25-50 ohm/cm and 200-250 μ sec, respectively. The UVM-2 apparatus, employed in the process of pulling the crystal under vacuum (10^{-4} mm Hg) in accordance with scheduled variations of the rate of pulling, utilizes a hydraulic raising mechanism which provides a smooth variation of the elevation rate from 0.05 to 8 mm/min in conjunction with the rotation of the crucible and the growing crystal. The ρ value of grown crystals 50 mm in diameter and 180 mm long deviated from the mean value of ~ 40 -60 ohm/cm, at a length of 100-150 mm, by 8.9-9.3% and was within the limits of error of measurement. It is pointed out that because of variations in conditions of crystallization only macrouniformity in the distribution of impurities can be achieved by this method. The variations were eliminated in another device which was designed for a process in which the pulling is accompanied by feeding of pure Ge into the melt. Prior to immersion into the melt, the feed ingot passes through a heating unit (composed of a quartz tube wound with W wire), while the ingot being pulled passes

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137-58-6-12122

Progress of Work on the Refining of Germanium (cont.)

through a water-cooled unit made of stainless steel. The process of pulling was carried out in an Ar atmosphere. The fact that the feed ingot and the growing crystal, which may rotate, were placed excentrically with respect to the axis of rotation of the crucible, contributed to a better mixing of the melt contained in the crucible. It was found that the longitudinal ρ fluctuations in single crystals obtained by this method did not exceed $\pm 5\%$, whereas in the case of a crystal produced in this apparatus without the employment of pure Ge the scatter amounted to $\pm 24\%$. Studies of crystals with a diameter of 50 mm have revealed that the ρ was sufficiently uniform throughout the cross section of the crystal and that the diffusion-path length of minority current carriers did not deviate by more than 10-15%. The authors express the opinion that intensive cooling of the growing crystal, which results in the achievement of a plane crystallization front, contributes to uniform distribution of electrical properties throughout the cross section of the crystal. Grown single crystals which exhibited nonuniform cross-sectional distribution of ρ were subjected to heat treatment in order to bring about a redistribution of ρ values; in the process the mean value of resistivity remained unaltered, while the diffusion-path length of the minority current carriers increased somewhat. 1. Germanium--Processing 2. Germanium--Crystallization 3. Single crystals--Growth 4. Single crystals--Properties I.S.

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ZEMSKOV, V.S.

137-58-5-9454

Translation from: Referativnyy zhurnal, Metallurgiya, 1958, Nr 5, p 92 (USSR)

AUTHORS: Petrov, D.A., Zemskov, V.S.

TITLE: Equipment and Methods of Growing Single Crystals of Semiconductors (Apparatura i metody vyrashchivaniya monokristallov poluprovodnikov)

PERIODICAL: V sb.: Rost kristallov. Moscow, AN SSSR, 1957, pp 262-272

ABSTRACT: An examination is made of the equipment and the results obtained in growing single crystals of Ge by the Chokhralskiy method and by drawing with feeding of additional substance. Drawing was performed by means of an apparatus consisting of 3 parts. At the bottom there was mounted a vacuum circulation chamber, connected with a system for evacuation. Water-cooled electrodes, to which were mounted a graphite slit heater and screens, were introduced into the chamber. Into the chamber from beneath there was introduced a motor-rotated rod carrying a crucible base. A quartz beaker with a viewing tap was placed atop the circulation chamber on rubber gaskets. The beaker was closed at the top by a water-cooled cover, to which was mounted

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137-58-5-9454

Equipment and Methods of Growing Single Crystals of Semiconductors

a mechanism for raising and rotating the seed-crystal-carrying finger passing through the cover into the chamber. The apparatus provided a vacuum of $\sim 10^{-4}$ mm Hg. The cover and lower chamber were connected by bolts for operation in a gas atmosphere. It is noted that in order for a single crystal to be grown the temperature of the melt should increase from its center toward the bottom and the walls of the crucible; some degree of overheating of the melt tends to increase the reliability of single crystal formation. To produce a single crystal, the seed crystal has to be fused along the entire interface between seed crystal and melt. Rotation of the crucible and the growing crystal is conducive to symmetrical growth of the crystal, but when the rate of crystal rotation is ~ 350 rpm, formation of a screw-shaped crystal is observed. There is a brief examination of the distribution of impurities in the drawing process. Changes in the apparatus to permit feeding of additional substance are described. The crystal being fed is passed through the heater, while the crystal being drawn passes through a cooler. This apparatus made it possible to obtain uniform longitudinal distribution of In^{114} along the single crystal drawn when a melt of pure Ge was fed in. This was verified by autoradiograph photometry and by resistance measurement (resistance variations of $\pm 5\%$). The importance of good mixing of the melt on drawing with feeding of additional substance is noted. 1. Single crystals--Growth 2. Germanium crystals--Growth Yu. Sh. Card 2/2 3. Indium isotopes (Radioactive)--Applications 4. Autoradiography--Applications 5. Semiconductors--Crystal structure

ZEMSKOV, V. S.

CHERKASHIN, Yevgeniy Yevgeniyevich; GLADYSHEVSKIY, Ye.I., dotsent, otv.
red.; ZEMSKOV, V.S., red.; SARANYUK, T.V., tekhred.

[Metric analysis of chemical equilibrium diagrams of systems
containing associated components] Metrika ravnovesnoi khimi-
cheskoi diagrammy sistem s assotsirovannymi komponentami.
Izd-vo L'vovskogo univ., 1958. 106 p. (MIRA 11:12)
(Systems (Chemistry))

LENEVICH, Lev Grigor'yevich, prof.; ZEMSKOV, V.S., red.; MALYAVKO,
A.V., tekhnred.

[Multiview method in descriptive geometry] Bezosnyi metod
v nachertatel'noi geometrii. Izd-vo L'vovskogo univ., 1958.
155 p. (MIRA 12:4)
(Geometry, Descriptive)

ZEMSKOV, V.S.

КЕКУА, М.О.; ЗЕМСОВ, В.С.

К вопросу получения равномерного распределения
примесей вольфрама при направлении роста монокристалла.

report submitted for the 5th Physical Chemical Conference on
Steel Production.

Moscow _ 30 JUL 64

24.7700

67293

SOV/180-59-4-26/48

AUTHORS: Zhurkin, B.G., Zemskov, V.S., Petrov, D.A. and
Suchkova, A.D. (Moscow)

TITLE: The Nature of the Quasi-Binary Germanium-Indium-Antimony
System

PERIODICAL: Izvestiya Akademii nauk SSSR, Otdeleniye tekhnicheskikh
nauk, Metallurgiya i toplivo, 1959, Nr 4, pp 156-158 (USSR)

ABSTRACT: Germanium with electron-type conduction and a specific
resistance 25 to 30 ohm/cm was used together with zone
refined antimony and indium. Crystals were pulled from
the melt. Results are given in Table 1. All the samples
had electron-type conductivity and samples with high InSb
content had a higher concentration of electrons than those
with low InSb content. The number of current carriers
varied from 1.2×10^{18} to $1.9 \times 10^{19}/\text{cm}^3$. The value for
fully compensated additions is $2.5 \times 10^{13}/\text{cm}^3$. Thus there
was an excess of Sb atoms. Experiments were carried out
using the same Ge:Sb ratio and increasing the In content.
Results are given in Table 2. With a ratio of In:Sb of
2.5 there is still electronic conduction very near to the
compensated alloy. With In:Sb = 4.4 there is hole-type
conduction. Microstructures were examined along the

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The Nature of the Quasi-Binary Germanium-Indium-Antimony System

length of the crystal pulled from a melt. A second phase appears (see Fig) which from microhardness tests corresponds to InSb. It does not appear, however, at temperatures greater than 650°C - the temperature of dissociation of InSb. The authors conclude that because of dissociation of InSb in fused germanium, the system does not possess the properties of a quasi-binary system. There are 1 figure, 2 tables and 8 references, 6 of which are Soviet and 2 English.

SUBMITTED: March 16, 1959

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67801

SOV/180-59-5-13/37

AUTHORS: Zhurkin, B.G., Zemskov, V.S., Petrov, D.A., and Suchkova, A.D. (Moscow)

TITLE: The Solubility of Indium and Antimony in Germanium and their Effect on some Electrical Properties of Germanium

PERIODICAL: Izvestiya Akademii nauk SSSR, Otdeleniye tekhnicheskikh nauk, Metallurgiya i toplivo, 1959, Nr 5, pp 86-90 (USSR)

ABSTRACT: Single crystals of germanium were pulled from melts doped with up to 80 wt % of indium or of antimony. [111] seeds were used; growth rate was 0.04 mm/min and the crystal was rotated at 140 rpm. Starting materials were: high purity germanium (25-30 ohm.cm N-type, mobility 3600 cm²/V.sec, diffusion length ~ 1.5-2 mm); indium showing spectrographic traces of Fe, Al, Cu, Ca, Ni and antimony of Cu, As, Pb, Au, Al and P. A pure graphite crucible fitted with a quartz sheathed thermocouple (Fig 1) held a charge of 10-12 g. The pulled ingots were 7-9 mm diameter and 8-10 mm long. These were cut in half lengthways. One half was studied metallographically for homogeneity while Hall effect specimens (7 x 3 x 1 mm) were cut from the other, close to the seed and perpendicular to the growth axis. Resistivity and Hall

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The Solubility of Indium and Antimony in Germanium and their
Effect on some Electrical Properties of Germanium

emf were measured with a potentiometer type PPTN-1 and a galvanometer type M-25/3. Resistivity measurements were $\pm 5\%$ but Hall measurements (3700 Oe field) for the higher impurity concentrations had greater errors, from 10-50%. In determining impurity concentrations from resistivity and Hall measurements complete ionization and degeneracy were assumed. The table shows equilibrium concentrations of indium and antimony in solid and liquid germanium at various temperatures (both wt % and at % values are given). The corresponding phase diagrams are plotted in Figs 3 and 4 (compositions in at %). Solid Ge containing $6.6 \cdot 10^{-2}$ at % In is in equilibrium with a melt containing 71.6 at % In at 620 °C, and solid germanium containing $7.2 \cdot 10^{-2}$ at % Sb with liquid containing 70.5 at % Sb at 693 °C. Extrapolation to the eutectic horizontals suggests maximum solid solubilities of $8 \cdot 10^{-2}$ at % In and about 0.1 at % Sb. No retrograde solid solubility was found for Sb. Fig 5 shows log-log plots (which are linear)

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6780x

SOV/180-59-5-13/37

The Solubility of Indium and Antimony in Germanium and their
Effect on some Electrical Properties of Germanium

of resistivity vs impurity concentration for Sb (1) and
In (2) doping. $2.5 \times 10^{-19} \text{ Sb/cm}^3$ gave $\sim 6 \cdot 10^{-4}$
ohm.cm, and $2 \cdot 10^{-19} \text{ In/cm}^3$ gave $2 \cdot 10^{-3}$ ohm.cm.

Fig 6 shows the corresponding variations in Hall
mobility; the plots for both holes and electrons
varying similarly. The results presented for In are
in good agreement with those in Ref 3.
There are 6 figures, 1 table and 14 references, of which
3 are Soviet, 10 English and 1 German.

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3/3

SUBMITTED: April 3, 1959

4

S/180/61/000/006/018/020
E073/E535

AUTHORS: Zemskov, V.S., Suchkova, A.D. and Wang Kuei-Hua
(Moscow)

TITLE: On the nature of heterogeneous equilibrium in the
system germanium-aluminium-antimony

PERIODICAL: Akademiya nauk SSSR. Izvestiya. Otdeleniye,
tekhnicheskikh nauk. Metallurgiya i toplivo,
no.6, 1961, 149-151

TEXT: In earlier work of the authors and their team
(Ref.1: Izv.AN SSSR, OTN, Metallurgiya i toplivo, 1959, No.4)
it was established that the Ge-InSb in the ternary Ge-In-Sb
does not represent the real equilibrium between the solid and
liquid phases during crystallization and, consequently, does not
have all the properties of a binary system. Therefore, the
necessity arose to verify the results obtained with the Ge-In-Sb
system on other systems and the Ge-Al-Sb system was chosen for
this purpose. The AlSb compounds are thermally more stable
than InSb compounds and, therefore, it was to be anticipated that
the behaviour of aluminium and antimony during crystallization of
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On the nature of heterogeneous ... S/180/61/000/006/018/020
E075/E535

germanium will differ from the behaviour of In and Sb. For investigating this system, the method of drawing the primary crystals of the solid solution from the appropriate melts containing various quantities of the alloying components was chosen. Measures were taken for creating equilibrium conditions of crystallization and for obtaining single-crystal specimens. The specimens were drawn at a rate of 0.045 mm/min; the seed was rotated at a speed of 100 to 120 r.p.m. and the crucible remained stationary. The germanium used had a specific resistance of 14 Ohm·cm and a diffusion length of 2.5 mm and an aluminium impurity content not exceeding 10^{-4} %; the antimony was grade CY-000 (SU-000). On the drawn crystals of the solid solutions the Hall constant and the specific resistance were measured. Two sections of the system containing 98 and 97 at.% Ge with various ratios of Al and Sb contents were investigated. The compositions of the initial melts of germanium from which primary crystals were produced are those given in Table 1. On the basis of the measured values of the Hall constant, the specific resistance of the solid solution crystals, of the determination

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On the nature of heterogeneous ... S/180/61/000/006/018/020
E073/E535

of the type of conductivity and of the calculated values of the concentrations of the current carriers, graphs were plotted of the changes in the electric parameters of the solid solutions as a function of the concentrations of Al and Sb in the conjugate liquid phases. It can be seen that crystals drawn from melts with an equi-atomic ratio of Al and Sb contents (curves a₁ and b₁) have a clearly pronounced hole-type conductivity. A change in the ratio of these elements until the Sb atoms predominate above the Al atoms by a factor of 20 does not lead to a change in the type of conductivity. If the ratio equals 22 to 25, the crystals will have a mixed-type conductivity and only if the content of Sb is over 25 times that of the Al content will there be a change to electron conductivity. An increase in the content of Sb in the liquid phase as compared to the content of Al leads to a change in the ratio of these elements in the solid phase towards full compensation, thereby reducing the number of current carriers, and the specific resistance will begin to increase. After compensation, the type of conductivity changes into electron conductivity and the specific resistance will again

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drop as a result of an increase in the quantity of Sb atoms above that of the Al atoms. In the system Ge-Al-Sb the observed changes in the electric characteristics are similar to those occurring in the system Ge-In-Sb; the section Ge-AlSb in the system Ge-Al-Sb is also not quasi-binary. If the ratio of the Sb atoms to the In atoms in the melt is 22 to 25, i.e. approaching the ratio of the coefficients of distribution of Al and Sb in binary systems with germanium (for Sb the coefficient of distribution at the given temperature is 2.9×10^{-3} and for Al it is 7×10^{-2}), the solid phase crystallizes with an equi-atomic ratio of Al and Sb atoms. The observed behaviour can be explained by dissociation of antimonides in the melting. Acknowledgments are expressed to B. G. Zhurkin for his assistance. There are 2 tables, 1 figure and 3 Soviet-bloc references.

SUBMITTED: May 18, 1961

Card 4/10 4

24.7700 (1035, 1043, 1055)

S/181/61/003/011/038/056
B104/B102

AUTHORS: Zhurkin, B. G., Zemskov, V. S., and Yurkina, K. V.

TITLE: Hall mobility of electrons in highly alloyed n-type germanium

PERIODICAL: Fizika tverdogo tela, v. 3, no. 11, 1961, 3509 - 3513

TEXT: The Hall mobility of electrons in n-type germanium monocrystals alloyed with antimony (up to $2.5 \cdot 10^{19} \text{ cm}^{-3}$) and arsenic (up to $6.0 \cdot 10^{19} \text{ cm}^{-3}$) was studied. Specimens cut from monocrystal ingots were used for measurements carried out at room temperature. The specimens had the dimensions $7 \cdot 3 \cdot 1 \text{ mm}$, the ingots had been produced by crystal pulling. The method of manufacturing strongly alloyed germanium was described in previous papers (B. G. Zhurkin et al., Izv. AN SSSR, OTN, no. 5, p. 86, 1959; V. S. Zemskov et al., Tezisy dokl. na konf. po udarnoy ionizatsii i tunnel'nomu efektu v poluprovodnikakh, Baku, 14 - 17, 1960). Electron-electron collisions were not taken into account because of degeneration in alloyed germanium. The Hall mobility was computed by the relation

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S/181/61/003/011/038/056

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Hall mobility of electrons in highly...

$\mu = R/\rho$, where R is the Hall constant and ρ the resistivity. The two latter quantities were measured with the use of a PPTN-1 (PPTN-1) voltmeter and an M25/3 (M25/3) galvanometer by a d-c compensation method. Magnetic fields of 3800-4200 oe were employed to measure R . The measuring error of ρ was $\pm 5\%$, that of the Hall-emf $\pm (10-20)\%$. The results showed that the Hall mobility of the electrons in germanium alloyed with antimony

($10^{18} - 10^{19} \text{ cm}^{-3}$) by far exceeds that of germanium of equal arsenic concentration. At an impurity concentration of about 10^{19} cm^{-3} the Hall mobility of electrons in germanium alloyed with antimony is almost twice that in germanium alloyed with arsenic. It is assumed that this difference is due to a change in effective electron mass m_n with the impurity. When electron scattering from impurity atoms and lattice vibrations was taken into account, an estimation of the effective masses $m_n(\text{Sb})$ and $m_n(\text{As})$ showed that within the concentration range of $5 \cdot 10^{17} - 2.5 \cdot 10^{19} \text{ cm}^{-3}$ $m_n(\text{Sb})$ was changed from 0.19 m to 0.30 m . In the range of arsenic concentration

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of $5 \cdot 10^{17} - 5 \cdot 10^{19} \text{ cm}^{-3}$, $m_n(\text{As})$ was changed from 0.195 m to 0.52 m. The authors thank B. M. Vul', L. V. Keldysh, and V. A. Chuyenkov for discussions. There are 1 figure, 2 tables, and 10 references: 3 Soviet and 7 non-Soviet. The three most recent references to English-language publications read as follows: W. Waring, D. Pitman, S. Steele, J. Appl. Phys., 29, no. 6, 1002, 1958; W. E. Baker, D. M. Compton, J. B. M. J. Res. and Develop., 4, no. 3, 275, 1960; M. Cardona, W. Paul, H. Brooks. Helv. phys. acta, 33, no. 5, p. 329, 1960.

ASSOCIATION: Institut metallurgii im. A. A. Baykova AN SSSR Moskva
(Institute of Metallurgy imeni A. A. Baykov AS USSR, Moscow)

SUBMITTED: March 3, 1961 (initially) July 3, 1961 (after revision)

Card 3/3

ZEMSKOV, V., polkovnik; BULATOV, A., polkovnik

Precarious period of a war and vigilance. Voenn. vest. 41 no.9:31-33
S '61. (MIRA 15:1)

(European War, 1914-1918) (World War, 1939-1945)
(United States--Militarism)

S/058/63/000/002/045/070
A062/A101

AUTHOR: Zemskov, V. S., Zhurkin, B. G., Suchkova, A. D., Yurkina, K. V.

TITLE: Production and properties of strongly alloyed germanium

PERIODICAL: Referativnyy zhurnal, Fizika, no. 2, 1963, 71, abstract 2E473
("Tr. Soveshchaniya po udarn. ionizatsii i tunnel'n. efektu v
poluprovodnikakh, 1960". Baku, AN AzerbSSR, 1962, 130 - 150)

TEXT: By the method of extracting the solid phase from a smelt with a large content of alloying admixture, single crystals of Ge were obtained with a concentration of Al up to $1.0 \cdot 10^{21} \text{ cm}^{-3}$, with a concentration of As up to $6.0 \cdot 10^{19} \text{ cm}^{-3}$, with a concentration of Sb up to $2.5 \cdot 10^{19} \text{ cm}^{-3}$ and with a concentration in In up to $2.0 \cdot 10^{19} \text{ cm}^{-3}$. It is established that there is an increase of solubility of In and Sb in Ge at a combined alloying, and this is explained on the basis of the electron-hole interaction in the solid phase. Applying the method of quantitative radiography and measuring the Hall effect made it possible to determine separately the concentration of In and Sb in the solid phase of Ge, while the data on the Hall mobility show an absence of neutral ion pairs $[\text{In}^- \text{Sb}^+]^0$. It was found that

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A062/A101

at room temperature the Hall mobility in Ge with Sb, in the range of Sb concentrations from $2 \cdot 10^{18}$ to $5 \cdot 10^{19}$, is about twice as high as the Hall mobility in Ge with As at equal concentrations of the admixtures.

[Abstracter's note: Complete translation]

Card 2/2

37735

S/180/62/000/002/014/018
E039/E435

10.8100

AUTHORS: Zemskov, V.S., Suchkova, A.D., Zhurkin, B.G.,
Wang Kuei-Hua (Moscow)

TITLE: The solubility of aluminium in germanium and the
influence of aluminium on some electrical properties of
germanium

PERIODICAL: Akademiya nauk SSSR. Izvestiya. Otdeleniye tekhnicheskikh
nauk. Metallurgiya i toplivo. no.2, 1962, 131-134

TEXT: The initial materials used were electron type germanium
with a specific resistance of 50 ohm.cm and aluminium purified by
zone melting containing Fe $< 7 \times 10^{-4}\%$, Mg $2 \times 10^{-4}\%$,
Si $4 \times 10^{-4}\%$, Cu $5 \times 10^{-5}\%$ and Zn $< 1 \times 10^{-4}\%$. Alloys were
prepared by the extraction method and all the investigated samples
were single phase. The region of solid solution was investigated
at 675°C. It is shown that the concentration of Al in solid
solution varied from 9.97×10^{-3} at.% at a concentration of
Al of 0.1 at.%, to 1.544 at.% at 46.2 at.% Al. A solidus curve
is plotted on which is included results from earlier papers.
Good agreement is obtained at 850°C but the earlier results
Card 1/2

The solubility of aluminium ...

S/180/62/000/002/014/018
EO39/E435

give rather lower values of Al concentration at temperatures less than 850°C. The concentration of current carriers was determined by measuring the Hall constant in a magnetic field of 4000 oersted and using the formula $n = A/Re$ where A is a coefficient, dependent on the diffusion mechanism of the current carriers, R is the Hall constant and e is the electronic charge. There are no accurate data for the change in A with concentration of acceptor atoms. However, estimated values are shown to give results within the limits of experimental error. The dependence of the specific resistance ρ on the concentration of Al atoms n is also determined and shown to fall on the same line as values for In and for In, Al, Ga obtained previously. ρ varies from 5×10^{-2} to 2×10^{-4} ohm cm for n varying from 10^{17} to 10^{21} cm⁻³. Values of the Hall mobility are calculated and shown to agree with earlier results. A weak dependence of μ on the concentration of acceptors is found at a concentration of Al atoms $> 10^{18}$ cm⁻³ which appears to be dependent on screening. There are 3 figures and 1 table.

SUBMITTED: May 31, 1961
Card 2/2

37736

S/180/62/000/002/015/018

E040/E535

18.1200

AUTHORS: Zemskov, V.S., Zhurkin, B.G. and Yurkina, K.V. (Moscow)

TITLE: The solubility of arsenic in germanium

PERIODICAL: Akademiya nauk SSSR. Izvestiya. Otdeleniye
tekhnicheskikh nauk. Metallurgiya i toplivo, no.2,
1962, 134-135

TEXT: Arsenic is one of the most commonly used alloying elements of germanium but, in spite of this, the available technical data for the solubility of arsenic in germanium are incomplete and often conflicting. The present investigation was carried out using a technique involving the extraction of samples from molten germanium solution containing various concentrations of arsenic and subsequent investigation of the specimens thus obtained, by X-ray structural and microscopic analyses, determination of the quantity of current carriers from measurements of the Hall constant and measurement of the specimen resistivity at room temperature. The starting materials for the investigations were germanium with the resistivity $\rho = 35-40$ ohm·cm and diffusion length of the minority current carriers of not less than 2-2.5 mm.
Card 1/2

The solubility of arsenic ...

S/180/62/000/002/015/018
E040/E535

The arsenic contained calcium and magnesium impurities in concentrations not exceeding 10^{-3} and $10^{-4}\%$, respectively. A partial phase composition diagram for the As-Ge system is constructed in semi-logarithmic coordinates in the temperature range 700 - 937°C and the solidus line is drawn in, together with the liquidus line quoted on the basis of data reported by H. St8hr and W. Klemm (Ref.6: Z.anorgan.und allgem.Chem., 1940, 244, p.205). It was established that the highest solubility of arsenic in germanium does not exceed 0.12 at.%. The above figure for the maximum solubility of arsenic in germanium-base solid solution agrees well with the value recently reported in the paper by P. L. Moody and A. J. Strauss (Ref.9: J. Electrochem.Soc., 1960, v.107, p.64). There are 1 figure and 1 table.

SUBMITTED: May 31, 1961

Card 2/2

S/076/62/036/009/003/011
B101/B102

AUTHORS: Zemskov, V. S., Suchkova, A. D., and Zhurkin, B. G. (Moscow)

TITLE: Study of the heterogeneous equilibrium in the system Ge-In-Sb

PERIODICAL: Zhurnal fizicheskoy khimii, v. 36, no. 9, 1962, 1914 - 1918

TEXT: The equilibrium between the solid and liquid phases was determined on three cross sections passing through the Ge-InSb cross section. The a - a cross section corresponded to a Ge content of 86.7 atom%, equilibrium temperature 860°C, the b - b cross section to a Ge content of 71 atom%, equilibrium temperature 812°C, and the c - c cross section to a Ge content of 41.2 atom%, equilibrium temperature 672°C. The concentration of admixtures (In, Sb) was determined in the Ge crystal pulled at 0.4 mm/min. n-type Ge was used, resistivity 30 ohm.cm, electron mobility 3600 cm/v.sec, diffusion length of minority carriers >2 mm. The crystals were examined radiographically, the number of carriers and the Hall constant were determined, and the macro- and microstructures were investigated. The Sb concentration was determined with the aid of Sb¹²⁴, the In concentration on the basis of the number of current carriers. Results:

Card 1/1

Study of the heterogeneous ...

S/076/62/036/009/003/011
B101/B102

(Fig.): Solid phases containing more Sb than In are in equilibrium with the Ge-InSb cross section. Thus, the Ge - InSb cross section is not a quasibinary system as it does not reproduce the true equilibrium between the solid and liquid phases. The increased solubility of In and Sb jointly present in Ge is explained by an electron - hole equilibrium in the solid phase, since the formation of $[\text{In}^+\text{Sb}^-]^0$ complexes is improbable at the experimental temperature. There is 1 figure.

ASSOCIATION: Institut metallurgii im. A. A. Baykova (Institute of Metallurgy imeni A. A. Baykov) ✓

SUBMITTED: January 7, 1961

Fig. Variation of the Sb and In contents in the solid phases as a function of their concentrations in the liquid phase. (a) = a - a cross section; (b) = b - b cross section; (c) = c - c cross section; (1) Sb in the solid phase; (2) In in the solid phase; atom% = atom%; dotted line = theoretical solubility.

Card 2/0 <

L 17996-63 EWP(q)/EWT(m)/BDS AFFTC/ASD/ESD-3 RM/JD
 ACCESSION NR: AP3001278 S/0181/63/005/006/1601/1604

AUTHORS: Zemskov, V. S.; Belaya, A. D.

TITLE: Interaction of aluminum and antimony during crystallization of germanium from melts containing these elements

SOURCE: Fizika tverdogo tela, v. 5, no. 6, 1963, 1601-1604

TOPIC TAGS: solid solution, Ge, Al, Sb, distribution coefficient

ABSTRACT: The authors have investigated the heterogeneous equilibrium in the three-member system Ge-Al-Sb. They studied solid solutions obtained from melts containing 97, 90, and 80 atomic % Ge with different Al-Sb ratios. They found qualitative support for earlier conclusions that a layer of Ge-AlSb does not reflect equilibrium between liquid and solid phases during crystallization of germanium melts according to the composition lying in this layer. The distribution coefficients of the elements show an interaction between atoms of the alloy components during changes in the ratios of these alloying elements in a melt.

SUB CODE: EL, FN

NO. 100: 001

Card 2/2

ACCESSION NR: AP4040989

S/0279/64/000/003/0154/0157

AUTHORS: Zemskov, V. S. (Moscow); Rozhdestvenskaya, V. V. (Moscow)

TITLE: Boron distribution during crystallization of solid solutions of germanium in silicon

SOURCE: AN SSSR. Izvestiya. Metallurgiya i gornoye delo, no. 3, 1964, 154-157

TOPIC TAGS: boron germanium, silicon, solid solution, semiconductor, doping, impurity content

ABSTRACT: The behavior of most doping components during crystallization of Ge and Si furnishes data on equilibrium coefficients of distribution depending on the concentration of the doping impurity in the melt. When the boron content is small in a melt ($3 \cdot 10^{18} \text{ cm}^{-3}$), its effective coefficient of distribution is found to be 17 during crystallization of Ge, 0.8 during crystallization of Si. But there have been no studies on the interaction between doping impurities and Ge and Si during crystallization of solid solutions of these semiconductors to give a continuous series of solid solutions. The authors now furnish results of studies on the distribution of boron during crystallization of Ge-Si solutions containing boron.

Card 1/2

ACCESSION NR: AP4040989

Their experiments show that when the Ge content in melts is increased to 30%, the effective coefficient of boron distribution increases from less than one to values near one and higher. This coefficient depends on the concentration of boron in the solid phase, and when the concentration changes from $6.2 \cdot 10^{18}$ to $4.4 \cdot 10^{20}$ atm/cm^3 , it is almost halved, which indicates a change in ionization conditions of boron atoms in the solid phase. "The authors thank N. Kh. Abrikosov for his constant interest in the work and V. P. Dmitriyev for his aid in measuring the electrical conductivity." Orig. art. has: 1 figure, 1 table, and 6 formulas.

ASSOCIATION: none

SUBMITTED: 09Sep63

ENCL: 00

SUB CODE: EC

NO REF SOV: 004

OTHER: 007

Card 2/2

ACCESSION NR: AP4043399

S/0181/64/006/008/2552/2554

AUTHORS: Zemskov, V. S.; Delaya, A. D.; Podkorytova, G. N.

TITLE: On the electric activity of tin in germanium

SOURCE: Fizika tverdogo tela, v. 6, no. 8, 1964, 2552-2554

TOPIC TAGS: tin, germanium, doping, liquid phase, solid phase, distribution statistics, single crystal, Hall effect, electroneutral molecule, impurity conduction, ionization energy

ABSTRACT: In view of the contradictory data on this subject, the authors attempt to ascertain the electric activity of tin in germanium by employing a theory of H. Reiss (J. Chem. Phys., v. 21, 1209, 1953), from which it follows that when the ionization conditions of the doping atoms change, an accompanying change takes place in the distribution coefficients of these elements between the liquid and solid phase. Thus, if tin is a donor, then its addi-

Card 1/4

ACCESSION NR: AP4043399

tion to germanium doped with an acceptor element should cause an increase in the distribution coefficient of both the tin and of the acceptor element. The acceptor element employed was gallium. The germanium single crystals were grown by a procedure described in detail elsewhere (FTT, v. 5, 1601 and 1100, 1963). The gallium concentration was determined by Hall-effect measurements. The obtained data on the concentration of gallium and tin in the solid phase were used to calculate the distribution coefficients and to plot the dependence of the distribution coefficients of gallium and tin on the ratio of these elements in the liquid phase. The results can be interpreted only by assuming that tin is not electrically neutral and that its donor nature is due to the unusual dependence of the distribution coefficient of gallium on its concentration in the melt. This explains also the disparity between the theoretically calculated and experimentally measured distribution coefficients of gallium, for the calculations were based on the assumption that the tin is electrically neutral. It is there-

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ACCESSION NR: AP4043399

fore concluded that tin serves as a donor impurity in germanium and that the ionization energy of the tin atoms is close to the ionization energy of the germanium. Orig. art. has: 1 figure.

ASSOCIATION: Institut metallurgii im. A. A. Baykova, Moscow
(Institute of Metallurgy)

SUBMITTED: 02Mar64

ENCL: 01

SUB CODE: SS

NR REF SOV: 004

OTHER: 006

Card 3/4

ZEMSKOV, V.S.; SUCHKOVA, A.D.; ZHURKIN, B.G.

Heterogeneous equilibrium in the system Ge - In - Sb. Zhur. fiz.
khim. 36 no.9:1914-1918 S '62. (MIRA 17:6)

1. Institut metallurgii imeni A.A. Boykova, Moskva.

GLAZOV, V.M.; ZEMSKOV, V.S.; ZHIRKIN, B.G.; SUCHKOVA, A.D.; LIU CHZHEN'-YUAN'
[Liu Chen-yuan]

Physicochemical analysis of systems $\text{Ge}(\text{Si}) - \text{A}^{\text{II}} - \text{B}^{\text{V}}$. Trudy
Inst. met. no.14:108-119 '63 (MJRA 17:8)

ZEMSKOV, V.S. (Moskva); RUZHITSKIY, V.V. (Moskva)

Distribution of boron during the crystallization of solid
solutions of germanium in silicon. Izv. AN SSSR Met. i gor. delo
no.3:154-158 My-Je'64 (MIRA 17:7)

ZEMSKOV, V.S.; BELAYA, A.D.; PODKORYTOVA, G.N.

Electric activity of tin in germanium. Fiz. tver. tela 6 no.8:
2552-2554 Ag '64. (MIRA 17:11)

1. Institut metallurgii imeni Baykova, Moskva.

ZEMSKOV, V.S.

Phosphorus vapor pressure over a silicon melt alloyed with phosphorus.
Izv. AN SSSR. Neorg. mat. 1 no.5:648-654 My '65. (MIRA 18:10)

1. Institut metallurgii imeni Baykova.